

# The Postulates of Quantum Mechanics.

A postulate is a hypothesis, if it agrees with experiment, then it can be taken as an axiom, a truth which we cannot prove. Quantum Mechanics is based upon a number of postulates; these are stated quite differently depending on the source, vary in actual number and order etc. depending on what book one reads. In these notes, I have used the four-postulate form found in Morrison, although I have added related information from other sources.

1 Every system can be described by a wavefunction (state function),  $\Psi(\mathbf{r},t)$  ( $\Psi(x,t)$  in 1D), which contains all accessible information about the system, including its spatial and temporal evolution.  $\Psi(\mathbf{r},t)$  will tell us everything!

2 The probability of finding a particle in a volume element  $dv$ , at  $\mathbf{r}$ , at any one time,  $P dv$ , is given by  $Pdv = |\Psi(\mathbf{r},t)|^2 dv = \Psi^*(\mathbf{r},t) \Psi(\mathbf{r},t) dv$  ( $Pdx = \Psi^*(x,t)\Psi(x,t)dx$  for finding our particle in the interval  $dx$  at  $x$  and  $t$ , in 1D). This means that  $\Psi^*(\mathbf{r},t)\Psi(\mathbf{r},t)$  is a probability density function for our system ( $\Psi^*(x,t)\Psi(x,t)$ ). Since the particle must exist somewhere in the universe, we obtain the normalisation condition, namely:

$$\int_{-\infty}^{\infty} \Psi^*(x,t)\Psi(x,t)dx = 1 \quad \text{for 1D}$$

3 Every observable (variable),  $q$ , is represented by an operator,  $\hat{Q}$ , and this operator is used to obtain information about the observable. e.g. momentum,  $p$ , operator  $\hat{P}$ ,  $\hat{P} = \frac{\hbar}{i} \frac{\partial}{\partial x}$ , position,  $x$ , operator  $\hat{X} = x$ , and so on... Combined with the wavefunction, they give rise to eigenvalue equations of the form:

$$\hat{Q}\Psi(x,t) = q\Psi(x,t)$$

...we can also obtain the average or expectation value for any observable:

$$\langle q \rangle = \int_{-\infty}^{\infty} \Psi^*(x,t)\hat{Q}\Psi(x,t)dx$$

Or, indeed, the uncertainty,  $\Delta q$ ,  $\Delta q = \sqrt{\langle q^2 \rangle - \langle q \rangle^2}$

4 The wavefunction or state function of an isolated system develops in time according to the Time Dependent Schrödinger Equation (TDSE), namely:

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V(x,t)\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

which may appear as:  $\hat{H}\Psi = i\hbar \frac{\partial \Psi}{\partial t}$  or even:  $\hat{H}\Psi = \hat{E}\Psi$

Since the wavefunctions in postulate 1 have to obey the TDSE in postulate 4, there are important conditions which apply to  $\Psi(x,t)$ . Amongst others, both  $\Psi(x,t)$  and its derivative,  $\frac{\partial \Psi(x,t)}{\partial x}$  must be continuous.