Nuclear Physics and Astrophysics

PHY-302

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Lecture 14 - Neutron Physics







Neutron Physics

Now switch attention to neutrons:

consider neutron applications in nuclear reactions & eg. crystallography will help us to understand fission & fusion in next part of course

- The neutron plays special role in study of nuclear forces
- Unaffected by coulomb forces \Rightarrow can penetrate nucleus at low energy $\sim 1 \text{ eV}$
- Initiate nuclear reactions
- Neutrons are difficult to detect no ionisation
- Neutrons are difficult to use in experiment cant focus / accelerate

Understanding and controlling these aspects of neutron physics is essential to the understanding of nuclear fission reactors



 $\begin{array}{ll} \label{eq:alpha} \alpha \mbox{-Beryllium source:} & Major source of neutrons $$^{9}Be has a loosley bound neutron collision with 5-6 MeV $$\alpha$ -particle can knock out neutron: $$^{4}He + $^{9}Be $$\rightarrow $^{12}C + n$$ neutron energy spectrum is very wide $$$ \end{tabular}$

Photoneutron Sources:

Similar to method above produces ~ monoenergetic neutron source $\gamma + {}^{9}Be \rightarrow {}^{8}Be+ n$

Spontaneous Fission:

fission of isotopes eg. ^{252}Cf produces 4n / fission neutrons energy range \sim 1-3 MeV

Nuclear Reactions:

Many nuclear reactions to choose from require accelerator to produce particle beam to initiate reaction neutrons can be tuned to be ~ monoenergetic e.g. ${}^{3}H + d \rightarrow {}^{4}He + n$

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Nuclear Reactors:

Neutron flux in fission reactor core is very high

Energy spectrum ~ I-7 MeV

Most are reduced to thermal temps in reactor

Some fast neutrons remain

Can be extracted by cutting hole in reactor shielding into lab

3



Neutron Detectors

neutrons produce no ionisation detectors must detect secondary events from neutron interactions: protons &-particles Y-rays fission products

Slow/thermal neutrons: detectors based on p⁺ emission or α -emission signals ¹⁰B often used: ¹⁰B + n \rightarrow ⁷Li^{*} + α Use ionisation chamber filled with BF₃ gas or coated with B metal Similar devices use ⁶Li, ³He

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Neutron Velocity Selection

neutrons velocities can be selected by mechanical means use rotating shutters made of Cd - high n absorption cross section rotating shutters only allow n of specific vel to pass through changing rotation speed, changes selection



5



Sources:

- neutron beams produced from variety of nuclear reactions
- impossible to accelerate them
- possible to slow them down via collisions with atoms moderation
 - E ~ 100 keV 10 MeV fast
 - slow E ~ I keV
 - epithermal E~IeV E ~ 0.025 eV
 - thermal

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Moderation:

As neutrons traverse matter many reactions occur Fast neutrons react via:

- beta decay to proton (n,p)
- α -emission in nuclear collision (n, α)
- n-emission in nuclear collision (n,3n)

Slow / thermal neutrons are mainly captured by nuclei

- neutron capture cross section usually dominated by one resonance

Consider neutron intensity loss traversing material thickness dx neutrons encounter ndx atoms per unit area n = #atoms per unit vol

 $\sigma_{_{t}}$ is absorption cross section then intensity loss is

 $\mathrm{d}I = -I \cdot \sigma_t \cdot n \cdot \mathrm{d}x$

$$I = I_0 e^{-\sigma_t nx}$$

Note: only accounts for neutrons at one energy



7



Neutron Collisions

Elastic collision: neutron (energy E, velocity v) on atom (mass A) E' is final neutron energy

E'/E

0.8

0.6

0.4

0.2

0

20 40

60 80 100

A=50

A=10

A=3

120 140 160

180

9

θ

$$\frac{E'}{E} = \frac{A^2 + 1 + 2A\cos\theta}{(A+1)^2}$$

For $\theta=0$ (no scattering) E=E' as expected For A=1 (hydrogen target) E'=0

Maximum energy loss is for head-on collision

$$\left[\frac{E'}{E}\right]_{min} = \left(\frac{A-1}{A+1}\right)^2$$

Typically E'/E varies from 1 to min

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Neutron Collisions

For E < 10 MeV scattering is ~ indep of θ For multiple collisions neutrons become increasingly less monoenergetic In limit of infinte collisions energy spectrum of neutrons like Maxwell distribution This is when neutron speed ~ thermal motion of atoms \Leftrightarrow thermal equilibrium !



Energy loss is largest for low A nuclei